Synthesis and Cytotoxicity Evaluation of Novel Indole Derivatives as Potential Anti-Cancer Agents

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<u>Abstract</u>

A series of bioisosteres of marine indole alkaloids (meridianins) was synthesized and the compounds were tested for their in vitro antiproliferative activity against HCT-116 cell-line. In the design of the targeted analogues, the 2-aminopyrimidine ring of merdianins was with 5-aminopyrazoles, pyrazolo[1,5-*a*]pyrimidines replaced and pyrazolo[3,4-*b*]pyridines. The cytotoxic screening of the synthesized compounds revealed that pyrazolo[1,5-*a*]pyrimidines (compounds 9c and **11a**) had the most potent cytotoxic activity with $IC_{50} = 0.31 \mu M$ and 0.34 **µM** respectively. Compounds **9c** and **11a** were further investigated for their kinase inhibitory potencies toward six kinases (CDK5/p25, CK1ð/ɛ, GSK- $3\alpha/\beta$, Dyrk1A, Erk2, and CLK1). The results showed that both compounds were highly potent inhibitors of GSK- $3\alpha/\beta$ and Erk2 enzymes.

Keywords: Indole, meridianins, cytotoxic activity, Erk2

Introduction

Cancer is the second leading cause of death across the globe which affect billions of people worldwide ¹. Recently, a number of essential antineoplastic agents that are in preclinical and clinical evaluation have been isolated from marine flora and fauna ^{2,3}. Marine invertebrates are considered as a potent source of different biologically active alkaloids ⁴.

The marine alkaloids meridianins A-G (Figure 1) are isolated from the south Atlantic tunicate *Aplidium meridianum*. Meridianins are indole derivatives substituted at position 3 with 2-aminopyrimidine ring. They inhibit many protein kinases such as cyclin dependent kinases, glycogen synthase kinase-3 and casein kinase-1 and display potent antitumor activity related to their enzyme inhibitory activity ⁵. Meridianin B and meridianin E are the most potent kinase inhibitors and cytotoxic agents. The SAR reveals that shifting the 2aminopyrimidine ring from position 3 to position 2 on the indole ring abolished the enzyme inhibitory activity. It is hypothesized that meridianins bind at the ATP kinase binding site *via* two H-bonds between the pyrimidine *N*-1 and the 2-amino group of the pyrimidine and both fix the molecule in the hinge region while the indole ring lays in the hydrophobic pocket ⁶⁻⁸.

With such a wealth of biological activity, meridianins represent promising lead structures to prepare more potent and selective kinase inhibitors and many analogues have been synthesized during the last two decades^{5,9}. Most of the published work focused on changing the substitution on the pyrimidine ring or the indole nitrogen. Thus, the pyrimidine ring was substituted at position 5 by aryl group or iodine ^{8,10}. However, the 5-arylpyrimidinyl substitution (for example compound **I**) was detrimental to the cytotoxic activity and enzyme inhibition of CDK5/p25, CK1ð/ɛ, GSK3α/β and Erk2, although their inhibitory activity of Dyrk1A was high. Dyrk1A enzyme is involved in Alzheimer's disease and Down syndrome.

On the other hand, the modifications reported on the indole ring involved substitution by alkyl group (*N*-methyl derivatives) ¹⁰, sulphonyl moieties ¹¹, or acyl groups ⁷. Both sulphonation and acylation of the indole nitrogen and the presence of 5-iodopyrimidine ring enhanced the inhibitory activity against Dyrk1A [4, 5, 6].

Other modifications of the meridianins structure included the isosteric replacement of the 2-aminopyrimidine ring by uracil¹² or pyridine ⁶ and the synthesis of bisindolylpyridines¹³, bisindolylpyrimidines (Compound **II**) or bisindolylpyridazines^{14,15}.

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However, no previous publications discussed the replacement of the pyrimidine ring by five membered ring or fused five and six membered rings.

In this work, indole analogues of meridianins were synthesized in which the pyrimidine ring was replaced by 5-aminopyrazole, pyrazolo[3,4-*b*]pyridine and pyrazolo[1,5-*a*]pyrimidine (Figure 2). Upon designing the compounds, the 7-amino group on the pyrazolo[1,5*a*]pyrimidines **4-9** was added to mimic the binding of the 2aminopyrimidine ring in meridianins. Different substitutions were added at position 5 (H, methyl, aryl) to examine the effect of different substitutions on the cytotoxic activity. Both 5,7-bisphenylpyrazolo[1,5*a*]pyrimidines **11a,b** and pyrazolo[3,4-*b*]pyridines **12,13** lack the amino group and only can act as H-bond acceptor. Besides, the latter compounds have large aromatic rings and can make hydrophobic interactions with the kinase binding site. The synthesized compounds were tested for their in vitro anti-proliferative activities toward HCT-116 cell-line. The kinase inhibitory potencies of the most potent compounds were examined toward six kinases (CDK5/p25, CK1ð/ε, GSK3α/β, Dyrk1A, CLK1 and Erk2).



Fig. 1: Merdianins and their reported analogues

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Fig. 2: Design of Merdianin analogues

Results and discussion

Chemistry

Schemes 1-3 outline the synthesis of the target compounds. The key precursor, 3-cyanoacetylindole (1) was prepared in excellent yield (91%) by heating a mixture of cyanoacetic acid and indole in acetic anhydride at 85 °C for 10 mins ¹⁶. In the present study, attempts to prepare compound 2a via refluxing one molar equivalent of compound 1 with three molar equivalent of hydrazine hydrate in ethanol as reported by Chiara *et al* ¹⁷ were unsuccessful and the starting compound was obtained unreacted. Trials to prepare compound 2b using phenyl hydrazine in ethanol and refluxing up to 13 hrs were also fruitless.

Nevertheless the targeted aminopyrazole derivatives were obtained by fusion of compound **1** either with 8 molar equivalents of hydrazine hydrate or 1 molar equivalent of phenyl hydrazine. (**Scheme 1**). The IR spectra of compounds **2a,b** indicated the appearance of NH₂ absorption bands at 3417-3325 cm⁻¹ and the disappearance of C=N and C=O absorption band. Moreover, the ¹H NMR spectrum of compound **2a** displayed two exchangeable singlet signals at δ 4.59 ppm and δ 11.25 ppm corresponding to NH₂ protons and NH proton of the pyrazole ring, respectively. While, the ¹H NMR spectrum of compound **2b** showed an exchangeable singlet signal at δ 5.36 ppm corresponding to NH₂ protons. Their ¹³C NMR spectra revealed the disappearance of aliphatic signals and the appearance of aromatic carbons at δ 88.0-154.2 ppm.

2a Compound was reacted with ethyl ethoxymethylenecyanoacetate 3a [18] in glacial acetic acid to afford the 7-aminopyrazolo[1,5-a]pyrimidine **4**. Whilst, the 7-oxo derivative **5** was afforded upon unexpectedly reacting compound with 2a ethoxymethylenecyanoacetate **3b**. This reaction proceeded via initial nucleophilic substitution of the ethoxy group of compound 3 by the exocyclic amino group of pyrazole, followed by intramolecular cyclization through nucleophilic addition of the pyrazole NH on either the cyano group to give compound **4** or the ester group to give compound **5**. [Fig. 3] Elemental analyses and spectral data were in favor of the proposed structures 4 and 5. Thus, the IR spectrum of compound 4 indicated the presence of ester C=O absorption band at 1678 cm⁻¹. While that of compound **5** showed cyano band at 2214 cm⁻¹ and C=O group at 1662 cm⁻¹. The ¹H NMR spectrum of compound **4** showed an exchangeable singlet signal at δ 8.61-8.84 ppm corresponding to $NH_{\rm 2}$ protons. Furthermore, triplet and quartet signals appeared at δ 1.34-1.37

ppm and δ 4.32-4.37 ppm assigned to ethyl protons. Both evidence proved that the cyclization occurred on the cyano group. The ¹³C NMR spectrum of compound **4** revealed the presence of aliphatic carbons at δ 60.9 ppm and δ 14.6 ppm, in addition to C=O carbon at δ 166.6 ppm. On the other hand, the ¹H NMR spectrum of compound **5** revealed the presence of a singlet signal at δ 2.51 ppm assigned to CH₃ protons and an exchangeable singlet signal at δ 13.37 ppm assigned to NH of the pyrimidine ring. Its ¹³C NMR spectrum showed the methyl carbon at δ 19.2 ppm, the C=N carbon at δ 116.3 ppm and C=O carbon at δ 158.1 ppm. Based on these findings, the cyclization was assumed to occur on the ester group.

Using similar approach, ethoxymethylenemalononitriles **6a,b** ^{20,21} and arylidenemalononitriles **8a-c** ^{22–24} were heated under reflux with compound **2a** in ethanol/pyridine mixture to give the pyrazolo[1,5-*a*]pyrimidines **7a,b** and **9a-c**, respectively (**Scheme 2**). The cyclocondensation reaction of 5-aminopyrazole **2a** chalcones **10a,b** in acidic medium afforded 5,7-bisphenyl derivatives **11a,b**.

Compound **2b** was reacted with acetylacetone or diethyl malonate in acetic acid to afford pyrazolo[3, 4-*b*]pyridines **12** and **13**, respectively (**Scheme 3**). Elemental analysis and spectral data were in favor of the proposed structures. The IR spectra indicated the disappearance of the amino group. The IR spectrum of compound **13** indicated the appearance of two C=O absorption bands at 1636 and 1635 cm^{-1.} Its ¹H NMR spectrum showed a singlet signal at δ 2.03 ppm assigned to -CO-<u>CH₂-</u> CO- and an exchangeable singlet signal appeared at δ 9.98 ppm corresponding to NH proton of the pyridine ring. While, the ¹³C NMR spectrum of compound **13** revealed the appearance of aliphatic protons at δ 24.9 ppm and two C=O carbons at δ 169.5 and 170.8 ppm.





Scheme 2: Synthesis of pyrazolo[1,5-*a*]pyrimidines



Fig. 3: Proposed mechanism for the synthesis of compounds 4 and 5

In vitro cytotoxic activity

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All the prepared compounds were tested for their *in vitro* cytotoxic activity on HCT-116 cell-line using MTT- cytotoxicity assay $^{25-30}$. Staurosporine was used as a reference compound and the results expressed as IC₅₀ in μ M are displayed in **Table 1** and represented graphically in **Fig. 4**.

The results indicated that some of the compounds showed cytotoxic activity comparable to Staurosporine. SAR study of the derivatives indicated that the 5-aminopyrazoles **2a,b** showed poor cytotoxic activity $IC_{50} = 48.25$ and 17.42 µM, respectively. This was expected since the 5-aminopyrazole ring cannot form efficient H-bond with the kinase enzyme as the 2-aminopyrimidine ring. Better activity was observed in case of pyrazolo[1,5-*a*]pyrimidine and pyrazolo[3,4-*b*]pyridine due to their larger size.

The most potent compounds were $9c~(\mathrm{IC}_{50}\text{=}~0.31~\mu\text{M})$ and $11a~(\mathrm{IC}_{50}\text{=}~0.34~\mu\text{M}).$

Table 1: In-vitro cytotoxicity against the HCT cell-line.

IC₅₀ (μM)
48.25 ± 1.2
17.47 ± 0.59
12.85 ± 0.4
30.06 ± 1.2
1.80 ± 0.03
91.12 ± 3.4
7.45 ± 0.18
5.80 ± 0.21
0.31 ± 0.01
0.34 ± 0.01
1.06 ± 0.03
2.34 ± 0.07
5.13 ± 0.18
4.1 ± 0.08
4.1 ± 0.08

Fig. 4: In-*vitro* cytotoxicity against the HCT-116 cell-line.



Kinase inhibitory activity

Compounds **9c** and **11a** which had the most potent cytotoxic activity were subjected to further *In vitro* Kinase inhibition assays. Their kinases inhibitory potencies were tested toward six kinases (CDK5/p25, CK1ð/ ϵ , GSK-3 α / β , Dyrk1A, Erk2, CLK1) and the results in terms of IC₅₀ were displayed in **Table 2** and represented graphically in **Figure 5**.

The results indicated that compounds **9c** and **11a** displayed submicromolar inhibition potencies toward most of the kinases tested but the significant inhibitory activity was against GSK-3 α / β and Erk2 enzymes [GSK-3 α / β (IC₅₀ = **0.196 µM** and **0.246 µM**, respectively) and Erk2 (IC₅₀ = **0.295 µM** and **0.376 µM**, respectively)]. Both GSK-3 α / β and Erk2 known to be involved in intestinal cell differentiation so had an effect on colon cancer cells ^{31–33}.

Table 2: Results of *in-vitro* kinase inhibition assays

Compoun	Kinase inhibition-IC ₅₀ in μM						
d No.	CDK5/p2	CK1ð/E	GSK-3α/β	Dyrk1A	Erk2	CLK1	
	5						
9c	0.863	0.633	0.196	52.36	0.295	12.76	
11a	1.281	0.893	0.246	0.827	0.376	0.777	

Fig. 5: Results of *in-vitro* kinase inhibition assays



Conclusion

In summary, series of novel indolylpyrazoles, indolylpyrazolo[1,5*a*]pyrimidines and indolylpyrazolo[3,4-*b*]pyridines were synthesized that revealed potent cytotoxic effect on HCT-116 cell-line. Compounds **9c** and **11a** showed the most potent cytotoxic activity with IC₅₀ = **0.31 µM** and **0.34 µM**, respectively. Besides, they exhibited effective inhibition of GSK-3 α / β (IC₅₀ = **0.196 µM** and **0.246 µM**, respectively) and Erk2 (IC₅₀ = **0.295 µM** and **0.376 µM**, respectively). Further work on meridianins scaffold is still needed to obtain more potent and selective kinase inhibitors.

Experimental

All the melting points were determined on Stuart apparatus and the values given are uncorrected. The IR spectra were determined using KBr discs on Shimadzu IR 435 spectrophotometer, microanalytical unit, Faculty of pharmacy, Cairo University and the values are represented in cm⁻¹.The ¹H NMR and ¹³C NMR spectra were carried out using a 400 MHz Bruker spectrophotometer, microanalytical unit, Faculty of pharmacy, Cairo University, Egypt, using TMS as internal standard. Chemical shift values are recorded in ppm on δ scale. The elemental analyses were carried out at the Regional center for Mycology and Biotechnology, Azher University, Egypt. The progress of the reactions was monitored by TLC using TLC sheets pre-coated with UV fluorescent silica gel Merck 60 F 254 and the sheets were visualized using UV lamp. Reagents purchased commercially were used without further purification. Solvents were dried using standard procedures.

General procedure for the synthesis of 3-(1*H*-indol-3-yl)pyrazol-5amines 2a,b.

A mixture of compound **1** (3.68 g, 0.02 mol), hydrazine hydrate (99%, 5.00 g, 5 mL, 0.10 mol) or phenyl hydrazine (2.16 g, 2 mL, 0.02 mol) was fused for 2 hrs. The reaction mixture was cooled. In case of compound **2a**, the fused product was triturated with water where greyish brown solid was formed. Compound **2b** was triturated with ethanol where yellow solid was formed. The solid was filtered, dried, washed with methanol (25 mL) and crystallized from ethanol.

3-(1H-Indol-3-yl)-1H-pyrazol-5-amine (2a)

Yield (71%), mp (219-221°C), IR (KBr) cm⁻¹: 3414, 3332 (NH₂), 3167, 3113 (NH), ¹H NMR (DMSO- d_6 , ppm) δ : 4.59 (s, 2H, NH₂, D₂O exchangeable), 5.74 (s, 1H, pyrazole Ar-H), 7.06-7.84 (m, 5H, Indole Ar-H), 11.25 (s, 1H, NH, D₂O exchangeable), 11.52 (s, 1H, NH, D₂O exchangeable), 11.52 (s, 1H, NH, D₂O exchangeable), 11.52 (s, 1H, NH, D₂O exchangeable), 112.2 (s, 1H, NH, D₂O, exchangeable), 112.2 (s, 1H, NH, D₂O, 121.9 (CDCl₃, ppm): 88.3 (CH-C(NH₂)-N-), 107.8, 112.2 (119.9, 120.2, 121.9, 123.1, 125.9, 136.7, 140.8, 154.2 (Ar-C), Anal. calcd for C₁₁H₁₀N₄: C,66.65, H, 5.08, N, 28.26. Found: C, 66.91, H, 5.17, N, 28.49.

3-(1*H*-Indol-3-yl)-1-phenyl-pyrazol-5-amine (2b)

Yield (67%), mp (209-211°C), IR (KBr) cm⁻¹: 3417, 3325 (NH₂), 3221 (NH), ¹H NMR (DMSO- d_6 , ppm) δ : 5.36 (s, 2H, NH₂, D₂O exchangeable), 5.85 (s, 1H, pyrazole Ar-H), 7.04-8.19 (m, 10H, Ar-H), 11.15 (s, 1H, NH, D₂O exchangeable), ¹³C NMR (CDCl₃, ppm): 88.0 (<u>CH</u>-C(NH₂)-N-N(C₆H₅)), 110.1, 111.9, 119.7, 121.6, 121.8, 122.6, 123.9, 125.5, 125.9, 129.5, 136.9, 140.21, 147.7, 148.0 (Ar-C), Anal. calcd for C₁₇H₁₄N₄: C, 74.43, H, 5.14, N, 20.42. Found: C, 74.70, H, 5.28, N, 20.13.

Ethyl 7-amino-2-(1*H*-indol-3-yl)pyrazolo[1,5-*a*]pyrimidine-6carboxylate (4).

A mixture of compound **2a** (0.5 g, 0.0025 mol) and compound **3a** (0.43 g, 0.0025 mol) in glacial acetic acid (20 mL) was heated under reflux for 8 hrs. The reaction was cooled and the separated solid was filtered, dried and crystallized from acetone.

Yield (51%), mp (233-235°C), IR (KBr) cm⁻¹: 3441, 3332 (NH₂), 3209 (NH), 2920, 2850 (CH-aliphatic), 1678 (C=O), ¹H NMR (DMSO-*d*₆, ppm) δ: 1.33-1.37 (t, 3H, <u>CH₃CH₂O-</u>, J=7.08 Hz), 4.31-4.36 (q, 2H, CH₃<u>CH₂O-</u>, J=7.08 Hz), 6.93 (s, 1H, pyrazole Ar-H), 7.15-8.59 (m, 6H,

Ar-H), 8.61 (s, 2H, NH₂, D₂O exchangeable), 11.56 (s, 1H, NH, D₂O exchangeable), ¹³C NMR (CDCl₃, ppm): 14.6 (<u>CH₃CH₂O-</u>), 60.9 (CH₃<u>CH₂O-</u>), 89.9, 93.8 (<u>pyrazole CH</u>), 108.6, 112.2, 120.7, 122.2, 122.5, 125.1, 127.0, 137.0, 149.0, 149.5, 150.8, 154.5 (Ar-C), 166.6 (C=O). Anal. Calcd for $C_{17}H_{15}N_5O_2$, C, 63.54, H, 4.71, N, 21.79. Found: C, 63.67, H, 4.94, N, 22.12.

2-(1*H*-Indol-3-yl)-5-methyl-7-oxo-4,7-dihydropyrazolo[1,5*a*]pyrimidine-6-carbonitrile (5).

A mixture of compound **2a** (0.5 g, 0.0025 mol) and compound **3b** (0.46 g, 0.0025 mol) in glacial acetic acid (20 mL) was heated under reflux for 15 hrs. Compound **5** gave a precipitate while hot. The reaction mixture was concentrated under reduced pressure and the separated solid was filtered, dried and crystallized from ethanol.

Yield (30%), mp (>300°C), IR (KBr) cm⁻¹: 3410, 3309 (2 NH), 2920, 2850 (CH-aliphatic) 2214 (CN), 1662 (C=O), ¹H NMR (DMSO-*d*₆, ppm) δ : 2.51(s, 3H, CH₃), 6.67 (s, 1H, pyrazole Ar-H), 7.14-8.32 (m, 5H, Ar-H), 11.56 (s, 1H, NH, D₂O exchangeable), 13.38 (s, 1H, NH, D₂O exchangeable), 13.38 (s, 1H, NH, D₂O exchangeable), 13.38 (s, 1H, NH, D₂O exchangeable), 116.3 (CN), 108.2, 112.2, 120.5, 121.7, 122.4, 125.2, 127.2, 137.0, 140.9, 152.6, 154.4, (Ar-C), 158.1 (C=O), Anal. Calcd for C₁₆H₁₁N₅O: C, 66.43, H, 3.83, N, 24.21. Found: C, 66.70, H, 4.05, N, 24.13.

General procedure for the synthesis of 7-amino-2-(1*H*-indol-3-yl)-5substitutedpyrazolo[1,5-*a*]pyrimidine-6-carbonitriles 7a, b. A mixture of compound **2a** (0.5 g, 0.0025 mol) and compound **6a** or compound **6b** (0.0025 mol) in ethanol (20 mL) was heated under reflux for 12-15 hrs in the presence of pyridine (1 mL) as a catalyst. Solid was precipitated on hot, the mixture was concentrated under reduced pressure. The separated solid was filtered, dried and crystallized from ethanol.

7-Amino-2-(1*H*-indol-3-yl)pyrazolo[1,5-*a*]pyrimidine-6-carbonitrile (7a)

Yield (58%), mp (^{*}300°C), IR (KBr) cm⁻¹: 3433, 3394 (NH₂), 3325 (NH), 2222 (CN), ¹H NMR (DMSO-*d*₆, ppm) δ: 6.96 (s, 1H, pyrazole Ar-H), 7.13-8.69 (m, 6H, Ar-H), 8.81 (s, 2H, NH₂, D₂O exchangeable), 11.55 (s, 1H, NH, D₂O exchangeable), ¹³C NMR (CDCl₃, ppm): 72.8, 94.2 (<u>pyrazole C</u>), 117.0 (CN), 108.6, 112.0, 120.5, 122.4, 122.9, 125.2, 127.2, 137.1, 149.3, 149.7, 151.5, 154.5 (Ar-C), Anal. Calcd for C₁₅H₁₀N₆: C, 65.68, H, 3.67, N, 30.64. Found: C, 65.37, H, 3.88, N, 30.91.

7-Amino-2-(1*H*-indol-3-yl)-5-methylpyrazolo[1,5-*a*]pyrimidine-6carbonitrile (7b)

Yield (76%), mp ($^{3}300^{\circ}$ C), IR (KBr) cm⁻¹: 3394, 3294 (NH₂), 3143 (NH), 2916 (CH-aliphatic), 2214 (CN), ¹H NMR (DMSO-*d*₆, ppm) δ : 2.49 (s, 3H, CH₃), 6.81 (s, 1H, pyrazole Ar-H), 7.12-8.65 (m, 5H, Ar-H), 8.67 (s, 2H, NH₂, D₂O exchangeable), 11.53 (s, 1H, NH, D₂O exchangeable), ¹³C NMR (CDCl₃, ppm): 24.1 (CH₃), 72.7, 93.2 (<u>pyrazole CH</u>), 117.1 (CN), 108.7, 112.0, 120.4, 122.3, 122.9, 125.2, 127.1, 137.1, 148.6, 149.7, 154.4, 159.5 (Ar-C), Anal. Calcd for C₁₆H₁₂N₆: C, 66.66, H, 4.20, N, 29.15. Found: C, 66.43, H, 4.39, N, 29.41.

General procedure for the synthesis of 7-amino-2-(1*H*-indol-3-yl)-5substitutedphenylpyrazolo[1,5-*a*]pyrimidine-6-carbonitriles 9a-c. A mixture of compound **2a** (0.5 g, 0.0025 mol) and compound **8a-c** (0.0025 mol) in absolute ethanol (20 mL) was heated under reflux in presence of pyridine (1 mL) as a catalyst. Solid was precipitated on hot, except for compound **9c** (precipitated after cooling). The mixture was concentrated under reduced pressure, the separated solid was filtered, dried and crystallized from ethanol.

7-Amino-5-(4-bromophenyl)-2-(1*H*-indol-3-yl)pyrazolo[1,5*a*]pyrimidine-6-carbonitrile (9a)

Reflux for 28 hrs, Yield (82%), mp (284-286°C), IR (KBr) cm⁻¹: 3437, 3402 (NH₂), 3294 (NH), 2202 (CN), ¹H NMR (DMSO-*d*₆, ppm) δ: 7.00 (s, 1H, pyrazole Ar-H), 7.14-8.72 (m, 9H, Ar-H), 8.84 (s, 2H, NH₂, D₂O exchangeable), 11.58 (s, 1H, NH, D₂O exchangeable), ¹³C NMR (CDCl₃, ppm): 71.5, 94.5 (<u>pyrazole CH</u>), 117.3 (CN), 108.6, 112.0, 120.5, 122.4, 122.9, 124.0, 125.2, 127.3, 131.1, 131.7, 137.15, 148.5, 150.7, 155.0, 157.8, 162.7 (Ar-C), Anal. Calcd for C₂₁H₁₃BrN₆: C, 58.76, H, 3.05, N, 19.58. Found: C, 58.90, H, 3.21, N, 19.63.

7-Amino-5-(4-fluorophenyl)-2-(1*H*-indol-3-yl)pyrazolo[1,5*a*]pyrimidine-6-carbonitrile (9b)

Reflux for 16 hrs, Yield (50%), mp (294-296°C), IR (KBr) cm⁻¹: 3444, 3387 (NH₂), 3309 (NH), 2206 (CN), ¹H NMR (DMSO- d_6 , ppm) δ : 6.99 (s, 1H, pyrazole Ar-H), 7.14-8.72 (m, 9H, Ar-H), 8.82 (s, 2H, NH₂, D₂O exchangeable), 11.58 (s, 1H, NH, D₂O exchangeable), ¹³C NMR (CDCl₃, ppm): 71.6, 94.4 (<u>pyrazole CH</u>), 117.4 (CN), 108.6, 112.0, 115.6, 120.5, 122.9, 125.2, 127.3, 131.3, 134.5, 137.1, 148.5, 150.7, 155.0, 157.9, 162.3, 164.7 (Ar-C), Anal. Calcd for C₂₁H₁₃FN₆: C, 68.47, H, 3.56, N, 22.81. Found: C, 68.90, H, 3.67, N, 22.72.

7-Amino-2-(1*H*-indol-3-yl)-5-(2-methoxyphenyl)pyrazolo[1,5*a*]pyrimidine-6-carbonitrile (9c)

Reflux for 36 hrs, Yield (35%), mp ($^{3}00^{\circ}$ C), IR (KBr) cm⁻¹: 3437, 3294, (NH₂), 3228 (NH), 2916, 2850 (C-H, aliphatic), 2229 (CN), ¹H NMR (DMSO-*d*₆, ppm) δ : 3.84 (s, 3H, OCH₃), 6.94 (s, 1H, pyrazole Ar-H), 7.07-8.69 (m, 9H, Ar-H), 8.71 (s, 2H, NH₂, D₂O exchangeable), 11.57 (s, 1H, NH, D₂O exchangeable), ¹³C NMR (CDCl₃, ppm): 55.9 (-OCH₃), 74.4, 94.2 (<u>pyrazole CH</u>), 116.8 (CN), 108.6, 112.0, 120.5, 120.8, 122.4, 122.8, 124.4, 125.2, 127.2, 127.7, 130.4, 131.5, 137.1, 148.7, 149.7, 154.6, 156.9, 158.5 (Ar-C), Anal. Calcd for C₂₂H₁₆N₆O: C, 69.46, H, 4.24, N, 22.09. Found: C, 69.72, H, 4.38, N, 22.37.

General procedure for the synthesis of 5,7-bis(4-halophenyl)-2-(1*H*-indol-3-yl)pyrazolo[1,5-*a*]pyrimidines 11a, b.

A mixture of compound **2a** (0.5 g, 0.0025 mol), and compound **10a** or compound **10b** (0.0025 mol) in glacial acetic acid (20 mL) was heated under reflux for 30 hrs. The mixture was cooled, the separated solid was filtered, dried and crystallized from ethanol.

5,7-Bis(4-bromophenyl)-2-(1*H*-Indol-3-yl)pyrazolo[1,5-*a*]pyrimidine (11a)

Yield (42%), mp ($^{>}$ 300°C), IR (KBr) cm⁻¹: 3402 (NH), ¹HNMR (DMSOd₆, ppm) δ : 6.38 (s, 1H, Pyrazole Ar-H), 6.40-8.23 (m, 14H, Ar-H), 11.11 (s, 1H, NH, D₂O exchangeable), ¹³C NMR (CDCl₃, ppm): 108.4, 109.7, 111.7, 114.8, 119.7, 120.8, 121.8, 122.6, 123.5, 126.4, 126.6, 129.7, 131.0, 131.7, 132.2, 136.1, 136.9, 138.27, 140.0, 145.1, 153.8, 154.4 (Ar-C), Anal. calcd for C₂₆H₁₆Br₂N₄: C, 57.38, H, 2.96, N, 10.29. Found: C, 57.43, H, 3.12, N, 10.43.

5,7-Bis(4-chlorophenyl)-2-(1*H*-indol-3-yl)pyrazolo[1,5-*a*]pyrimidine (11b)

Yield (55%), mp ($^{>}$ 300°C), IR (KBr) cm⁻¹: 3398 (NH), ¹HNMR (DMSOd₆, ppm) δ : 6.96 (s, 1H, Pyrazole Ar-H), 6.99-8.33 (m, 14H, Ar-H), 11.10 (s, 1H, NH, D₂O exchangeable), ¹³C NMR (CDCl₃, ppm): 108.4, 109.8, 111.7, 115.0, 119.7, 120.8, 121.8, 126.4, 126.6, 128.1, 129.3, 129.5, 131.5, 133.9, 134.8, 136.0, 136.6, 137.92, 140.0, 145.1, 153.7, 154.3 (Ar-C), Anal. calcd for C₂₆H₁₆Cl₂N₄: C, 68.58, H, 3.54, N, 12.30. Found: C, 68.23, H, 3.62, N, 12.49.

3-(1*H*-Indol-3-yl)-4,6-dimethyl-1-phenyl-1*H*-pyrazolo[3,4-*b*]pyridine (12).

A mixture of compound **2b** (0.69 g, 0.0025 mol) and acetylacetone (0.25 g, 0.0025 mol) in glacial acetic acid (20 mL) was heated under reflux for 16 hr. The reaction was concentrated under reduced pressure, then few drops of water was added, the obtained emulsion was left to settle in the fridge overnight. The separated solid was filtered, dried and crystallized from ethanol.

Yield (43%), mp (198-200°C), IR (KBr) cm⁻¹: 3398 (NH), 2954, 2850 (CH-aliphatic), ¹H NMR (DMSO-d₆, ppm) δ : 2.51 (s, 3H, CH₃), 2.63 (s, 3H, CH₃), 7.05-8.40 (m, 11H, Ar-H), 11.53 (s, 1H, NH, D2O exchangeable), ¹³C NMR (CDCl₃, ppm): 19.8, 24.9 (CH₃), 108.1, 112.2, 114.6, 120.0, 120.2, 120.4, 120.9, 122.2, 125.8, 127.1, 127.4, 129.5, 136.5, 140.0, 141.1, 143.8, 151.2, 158.7 (Ar-C). Anal. calcd for C₂₂H₁₈N₄: C, 78.08, H, 5.36, N 16.56. Found: C, 78.35, H, 5.48, N, 16.67.

3-(1*H*-Indol-3-yl)-1-phenyl-1*H*-pyrazolo[3,4-*b*]pyridine-4,6(5*H*,7*H*)dione (13).

A mixture of compound **2b** (0.69 g, 0.0025 mol) and diethyl malonate (0.40 g, 0.0025 mol) in glacial acetic acid (20 mL) was heated under reflux for 18 hr. The reaction was concentrated under reduced pressure, then few drops of water was added, the obtained emulsion was left to settle in the fridge overnight. The separated solid was filtered, dried and crystallized from ethanol. Yield (47%), mp (207-209°C), IR (KBr) cm⁻¹: 3221, 3190 (NH), 2920, 2850 (CH-aliphatic), 1634, 1635 (2 C=O), ¹H NMR (DMSO-*d*₆, ppm) δ : 2.03 (s, 2H, CO-<u>CH</u>₂-CO), 6.75-8.21 (m, 10H, Ar-H), 9.98 (s, 1H, NH, D₂O exchangeable), 11.31 (s, 1H, NH, D₂O exchangeable), ¹³C NMR (CDCl₃, ppm): 24.9 (CH₂), 79.6, 100.3, 112.0, 115.8, 118.0, 119.9, 121.4, 122.0, 123.5, 129.6, 136.8, 139.5, 149.2, 155.7, 157.6, 169.5 (C=O), 170.8 (C=O), Anal. Calcd for C₂₀H₁₄N₄O₂: C, 70.17, H, 4.12, N, 16.37. Found; C, 70.35, H, 4.29, N, 16.70.

In vitro cytotoxic screening

The synthesized merdianin analogues were evaluated for their cytotoxic activity using MTT- cytotoxicity assay on HCT-116 (colon carcinoma cell-line). The test compounds were dissolved in DMSO, and Staurosporine was used as a reference.

Cell Line cells were obtained from American Type Culture Collection, cells were cultured using DMEM (Invitrogen/Life Technologies) supplemented with 10% FBS (Hyclone), 10 ug/ml of insulin (Sigma), and 1% penicillin-streptomycin. All other chemicals and reagents were obtained from Sigma-Aldrich, or Invitrogen.

Cell culture protocol. The culture medium was removed to a centrifuge tube, and the cell layer was briefly rinsed with 0.25% (w/v) Trypsin 0.53 mM EDTA solution, to remove all traces of serum which contains Trypsin inhibitor. Then 2-3 L of the Trypsin-EDTA solution was added to the flask and cells were observed under an inverted microscope until the cell layer was dispersed. After that, 6-8 L of the complete growth medium was added, and cells were aspirated gently by gently pipetting. The cell suspension was transferred to the centrifuge tube with the medium and cells from the first step, and it was centrifuged at 125 xg for 5 to 10 min, and the supernatant was discarded. The cell pellet was resuspended in a fresh growth medium, and the appropriate aliquots of cell suspension were added to new culture vessels. Finally, the cultures were incubated at 37 for 24 h.

<u>MTT – Cytotoxicity assay protocol:</u>

Plate cells (cell density 1.2-1.810,000 cells/well) in a volume of 100 L complete growth medium 100 L of the tested compound per well in a 96-well plate were incubated for 24 h before the MTT assay. A series of concentrations (0.01, 0.1, 1, 10, 100M) of the panel of tested compounds were added to cells. After 24 h incubation, MTT stain (10L per 100L of the culture medium) was added, and the cells were incubated for additional 4h. The formed formazan crystals were dissolved in MTT Solubilization Solution [Sigma catalogue no. M-8910], and gently mixed in a gyratory shaker. The cultures were pipette up and down to completely dissolve the MTT formazan crystals. The absorbance of each cell was measured at a wavelength of 450 nm by BIOLINE EIIZA reader. The relation between surviving fraction and compound concentration was plotted and IC₅₀ values expressed in M as the mean values of triplicate

wells from at least three experiments were calculated and reported as the mean standard error.

Kinase inhibitory activity

The kinase inhibitory activity of the synthesized compounds was determined using Human proto-oncogene serine/threonine-protein kinase **GSK-3** α / β ELISA Kit (catalog #V9361), **Dyrk1A** (catalog #D09-10G), **CK1ð**/ ϵ (catalog #CS0400), **CLK1** (catalog #V4057), **Erk2** (catalog #V9291) and **CDK5**/**p25** (catalog #C33-10G-10). All the compounds were tested for their inhibitory activity against kinases at 0.2, 1, 5 and 25 µM. The results were displayed in terms of IC₅₀. **Table 2** and **Figure 5** show the obtained results.

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